

**ENVIRONMENT & ENERGY / ENVIRONMENTAL REMEDIATION**

**PROJECT:** Environmental Remediation Science & Technology: *Re-oxidation of Redox Sensitive Contaminants Immobilized by Strong Reductants*

**CLIENT:** U.S. Department of Energy  
**PRINCIPAL INVESTIGATOR:** Dr. Leonel Lagos  
**LOCATION:** Richland, Washington

**Description:**

FIU's Applied Research Center (ARC) is providing experimental support to Pacific Northwest National Lab's (PNNL's) for potential *in situ* treatment technologies for vadose, groundwater, and perched water zones located within the 200 Area at the Hanford Site.

The perched water zone located beneath the 200-DV-1 Operable Unit at Hanford presents a challenge due to elevated concentrations of uranium (U) and technetium (<sup>99</sup>Tc). There is a potential for these contaminants to infiltrate into the underlying aquifer, exacerbating the existing groundwater contamination issue, which includes elevated levels of <sup>99</sup>Tc and nitrate (NO<sub>3</sub><sup>-</sup>).

This research focuses on the application of strong reductants, namely zero valent iron (ZVI), sulfur-modified iron (SMI), and calcium polysulfide (CPS), to sequester the co-located contaminants. These reductants have shown promise in immobilizing U and Tc by reducing them to lower oxidation states that are less mobile and soluble.

It is important to evaluate the potential re-oxidation of the previously immobilized contaminants under aerobic (oxidizing) conditions. Both Tc(VII) and U(VI) can undergo re-oxidation and re-mobilization when exposed to oxidizing conditions. Therefore, this research assesses the effectiveness of the applied reductants in preventing re-oxidation and ensuring long-term immobilization of the contaminants.

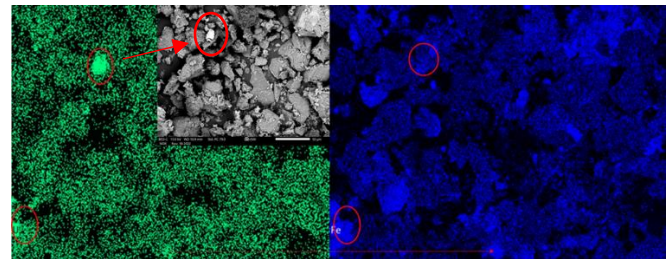
**Benefits:**

This research evaluates re-oxidation behavior and kinetics of <sup>99</sup>Tc in the presence of other co-

contaminants (e.g., <sup>238</sup>U and NO<sub>3</sub><sup>-</sup>) after initial immobilization via reduction. The results from these experiments will be used to determine the extent of Tc re-oxidation in the presence of co-contaminants, uranium, and nitrate, to be able to evaluate the effects of different reactants and their concentrations and provide insights for remedial actions.

**Accomplishments:**

- FIU investigated re-oxidation behavior of perched and groundwater contaminants, such as <sup>99</sup>Tc(VII), in the presence of <sup>238</sup>U and NO<sub>3</sub><sup>-</sup>, that have been initially reduced by strong reductants such as 0.1% and 1% ZVI and SMI in batch experiments under initial anaerobic conditions followed by aerobic conditions.



**Figure 1. EDS maps of PW sample treated with 1% SMI sacrificed after Phase 1 showing alignment of U and Fe.**

- Experiments determined that under anaerobic conditions, both ZVI and SMI exhibited effective reduction of all contaminants. In aerobic conditions, ZVI demonstrated slightly better resistance to re-oxidation compared to SMI.
- In sediment-free perched water (PW) samples treated with 1% MSI, there was a higher degree of resistance to re-oxidation observed for U and Tc compared to ZVI. The concentration of re-oxidized U in the sediment-free PW solution remained high, suggesting the need for additional methods to effectively sequester U.
- These findings highlight the challenges associated with the re-oxidation of U and the necessity for additional sequestration methods indicative of the complexity of the remediation process.

**ABOUT**

Since 1995, the Applied Research Center (ARC) at Florida International University (FIU) has provided critical support to the Department of Energy's Office of Environmental Management (DOE-EM) mission of accelerated risk reduction and cleanup of the environmental legacy of the nation's nuclear weapons program. ARC's applied research is performed under the DOE-FIU Cooperative Agreement (under Contract # DE-EM0000598) and provides technical support to DOE EM in the area of environmental remediation and STEM workforce development and training.

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